**Original Research Article**

# Effect of Temperature on Thermodynamic Properties of Glycine, Histidine, and Valine in Aqueous Solutions of Nortriptyline

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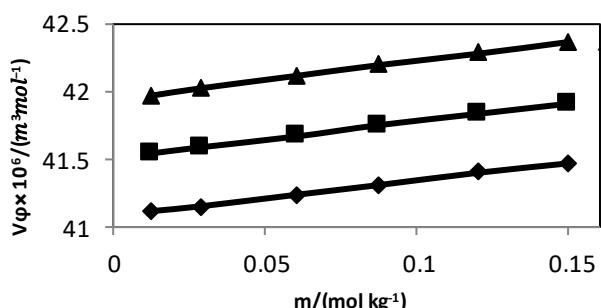
Valine

Nortriptyline

## ABSTRACT

Thermodynamic properties such as apparent molar volume,  $\nu\varphi$  and apparent molar isentropic compressibility,  $k_{\omega s}$  for aqueous solutions of glycine, histidine, and valine in nortriptyline corresponding measured their density as well as speed of sound at  $T = (305.15, 310.15, \text{ and } 315.15)$  K were computed. The obtained quantities were discussed in terms of interactions between drug-water and amino acids which was due to specification structural and upon the situation of nortriptyline molecules.

## GRAPHICAL ABSTRACT



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## 1. INTRODUCTION

Physico-chemical quantities of glycine, histidine, and valine in aqueous solutions lead to understanding interactions between unlike molecules [1-4]. On the other hand, the behavior of amino acids in antidepressant drug recently is of high significance [5-9]. In this project, the drug nortriptyline was used to discuss the interactions with amino acids. For this purpose, first densities and speeds of sound were measured for pure components as well as for binary mixtures with different molalities at  $T = (305.15, 310.15, \text{ and } 315.15)$  K. Consequently, the apparent molar volume such as apparent molar volume,  $\nu\omega$ , and apparent molar isentropic compressibility,  $k_{\omega s}$  were also calculated. Compared to the other tricyclic anti-depressant, nortriptyline is less toxic and displays less drug interactions. It undergoes hepatic metabolism and hydroxylation followed by conjugation with glucuronic acid [10-12].

## 2. EXPERIMENTAL

The amino acids and nortriptyline with purities of  $>0.99$  were used without further purification. The different molalities of solutions from (0.0006-0.004)  $m$  of nortriptyline were made. Densities and sound velocities were measured by Anton Paar DSA 5000 M densitometer. [Table 1](#) presents the purities outlines chemical specifications, purities, and supplier.

## 3. RESULTS AND DISCUSSION

The quantity  $\nu\omega$ , of solutions were calculated as:

$$V_\varphi = \frac{1000(\rho_0 - \rho)}{m\rho\rho_0} + \frac{M}{\rho} \quad (1)$$

Where,  $M$  is molar mass and  $m$  is molality.  $\rho_0$  is density of pure solvent and  $\rho$  is density of solution. The obtained  $\nu\omega$  values for various molality at  $T = (305.15, 310.15 \text{ and } 315.15)$  K are reported in [Table 2](#). The  $\nu\omega$ , values are positive and become more positive with enhancing molality of amino acid and temperatures in

nortriptyline solutions. The sign of  $\nu\omega$  values is due to interactions between unlike molecules and increases with glycine to histidine at different temperatures. The  $\nu\omega$ , values of histidine, valine, and glycine in water and in aqueous nortriptyline solutions at different molality and at  $T = (305.15, 310.15, \text{ and } 315.15)$  K are presented in [Table 3](#), and displayed in [Figures 1-3](#). The values apparent molar volume at dilute solutions  $V_\omega^\circ$  and slope of plot  $V_\omega$  vs molality as  $S_v$ , are obtained as:

$$\nu\omega = V_\varphi^0 + S_v m \quad (2)$$

The weaker interactions between solutes rather than solutions made positive lower values of  $S_v$ , relative to  $V_\omega^\circ$  the computed values of  $V_\omega^\circ$  and  $S_v$ , are listed in [Table 4](#) [13-16].

To calculate isentropic compressibility,  $k_s$  the Newton-Laplace equation was used:

$$K_s = \frac{1}{\rho u^2} \quad (3)$$

$u$  and  $\rho$  are speed of sound and density, respectively. Apparent molar isentropic compressibility was calculated as follow:

$$K_{\varphi,s} = \frac{1000(K_s\rho_0 - K_s^0\rho)}{m\rho\rho_0} + \frac{MK_s}{\rho} \quad (4)$$

Where,  $\rho$  and  $\rho_0$  show densities of the solution and the solvent. The resultant  $k_{\omega s}$  is indicated in [Table 3](#) and [Figures 4-6](#). Results showed that  $k_{\omega s}$  were negative for all systems which indicate that water molecules around the solute are less compressible than those in the bulk. The limiting apparent molar isentropic compressibility  $k_{\omega s}^0$  was calculated as follow:

$$K_{\varphi,S} = K_{\varphi,s}^0 + S_k m \quad (5)$$

$S_k$ , shows output solute-solute interactions and  $k_{\omega s}^0$ , indicates solute-solvent interactions. The

computed values of  $\mathbf{k}_{\phi,s}$  and  $S_k$  are presented in [Table 4](#) [17–21].

The  $V_{\phi^*}$  is transfer volume of amino acids from water to nortriptyline solutions and is given as follow:

$$V_{\phi,tr}^0 = V_{\phi,aqueous-nortriptyline}^0 - V_{\phi,water}^0 \quad (6)$$

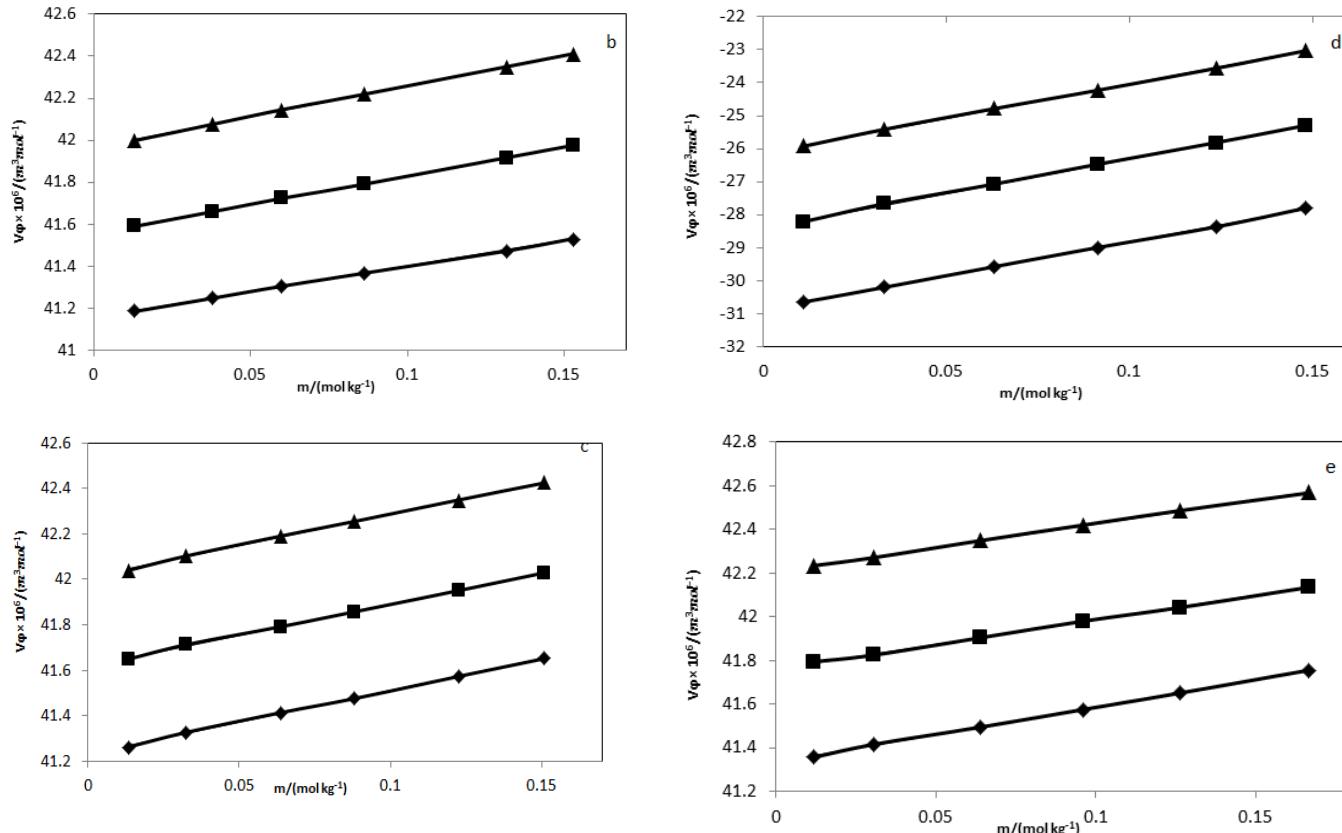
The values  $V_{\phi^*}$  is for histidine, valine, and glycine in water and transfer compressibility,  $\mathbf{k}_{\phi,s,tr}$  of amino acids from water to solutions was obtained as follow [\[22–24\]](#):

$$K_{\phi,s,tr}^0 = K_{\phi,s,aqueous-nortriptyline}^0 - K_{\phi,s,water}^0 \quad (7)$$

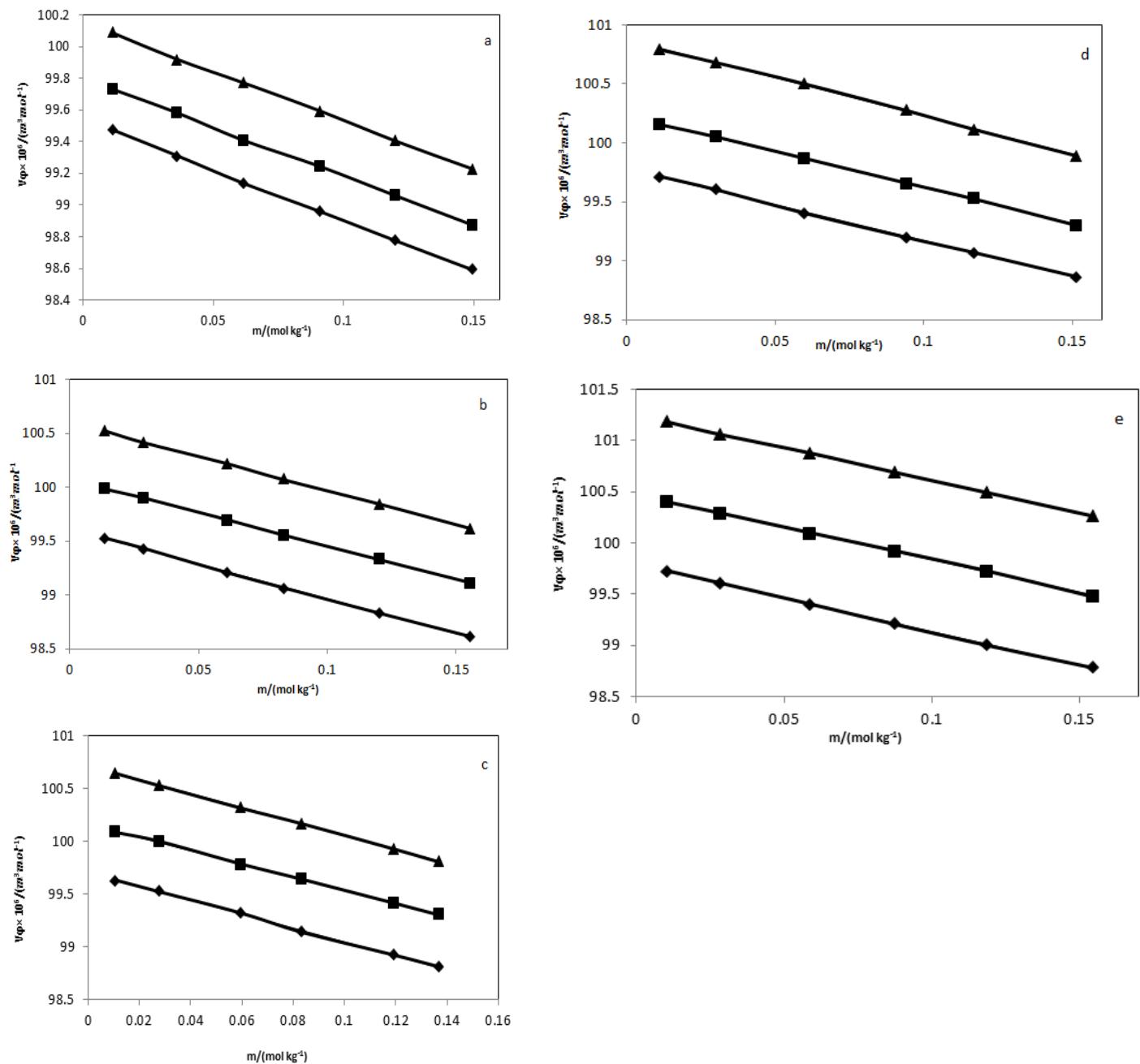
The computed values of  $\mathbf{k}_{\phi,s,tr}$  and  $V_{\phi,tr}$  are reported in [Table 5](#).

#### 4. Conclusion

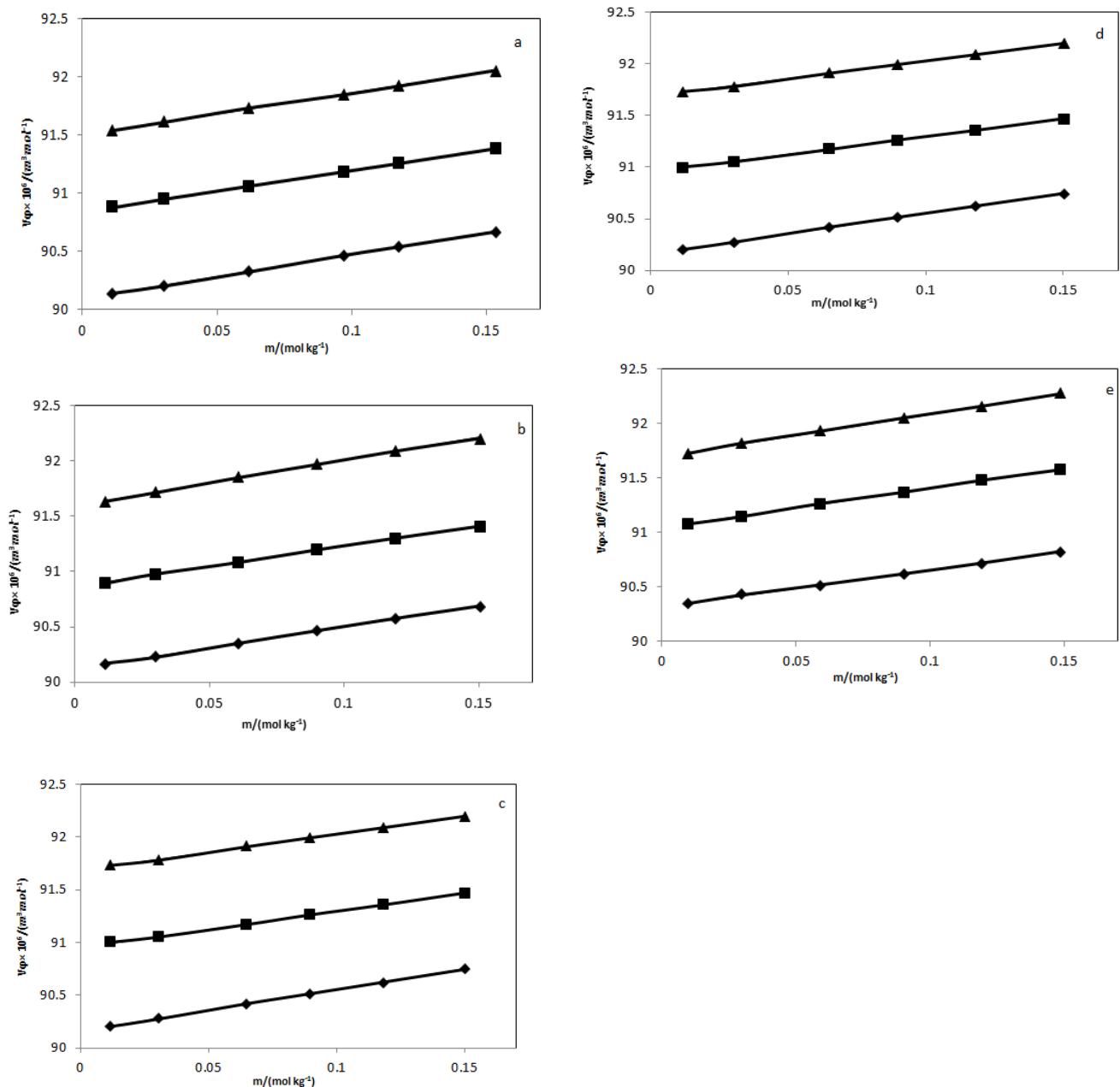
The density and speed of sound for amino acids in aqueous solutions of nortriptyline drug at  $T = (305.15, 310.15, \text{ and } 315.15) \text{ K}$  were experimentally measured. From the experimental data, various parameters  $\mathbf{v}\omega$ ,  $V_{\phi^*}$ ,  $\mathbf{k}_{\phi,s}$ ,  $\mathbf{k}_{\phi,s,tr}$ , and  $V_{\phi,tr}$  were calculated. The different parameters showed that there are strong interactions between amino acids and drug molecules. This strength of solute-solvent interaction enhances with increase in molar mass of amino acids and molalities.



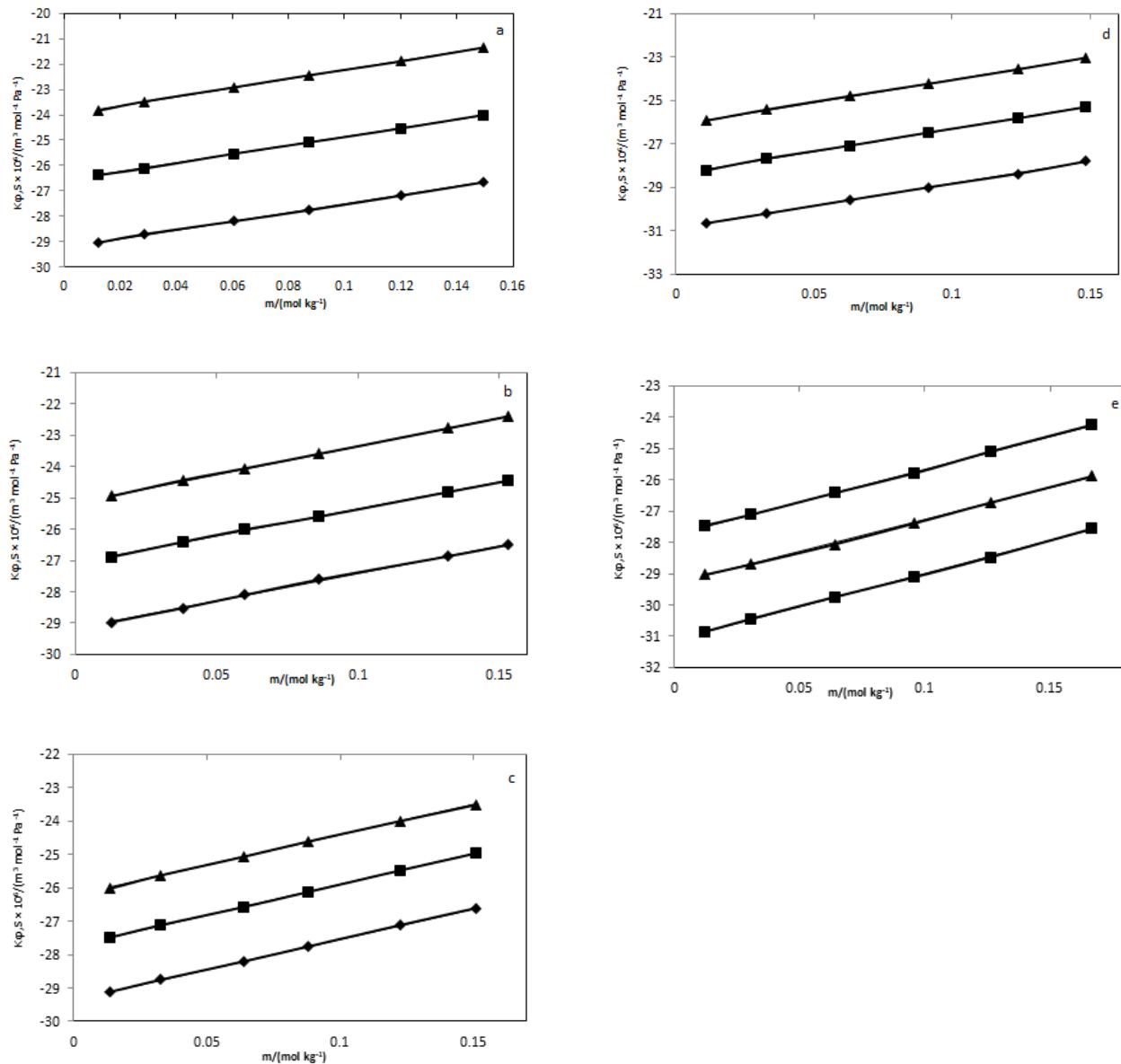
**Fig 1.** Plot of apparent molar volume  $\mathbf{v}\omega$ , vs. molality , $m$  of (a) glycine water, (b) glycine in 0.0006 m aqueous nortriptyline, (c) glycine in 0.001 m aqueous nortriptyline, (d) glycine in 0.002 m aqueous nortriptyline, and (e) glycine in 0.004 m aqueous nortriptyline at temperatures: 305.15 K( $\emptyset$ ), 310.15 K( $\square$ ), and 315.15 K( $\Delta$ )



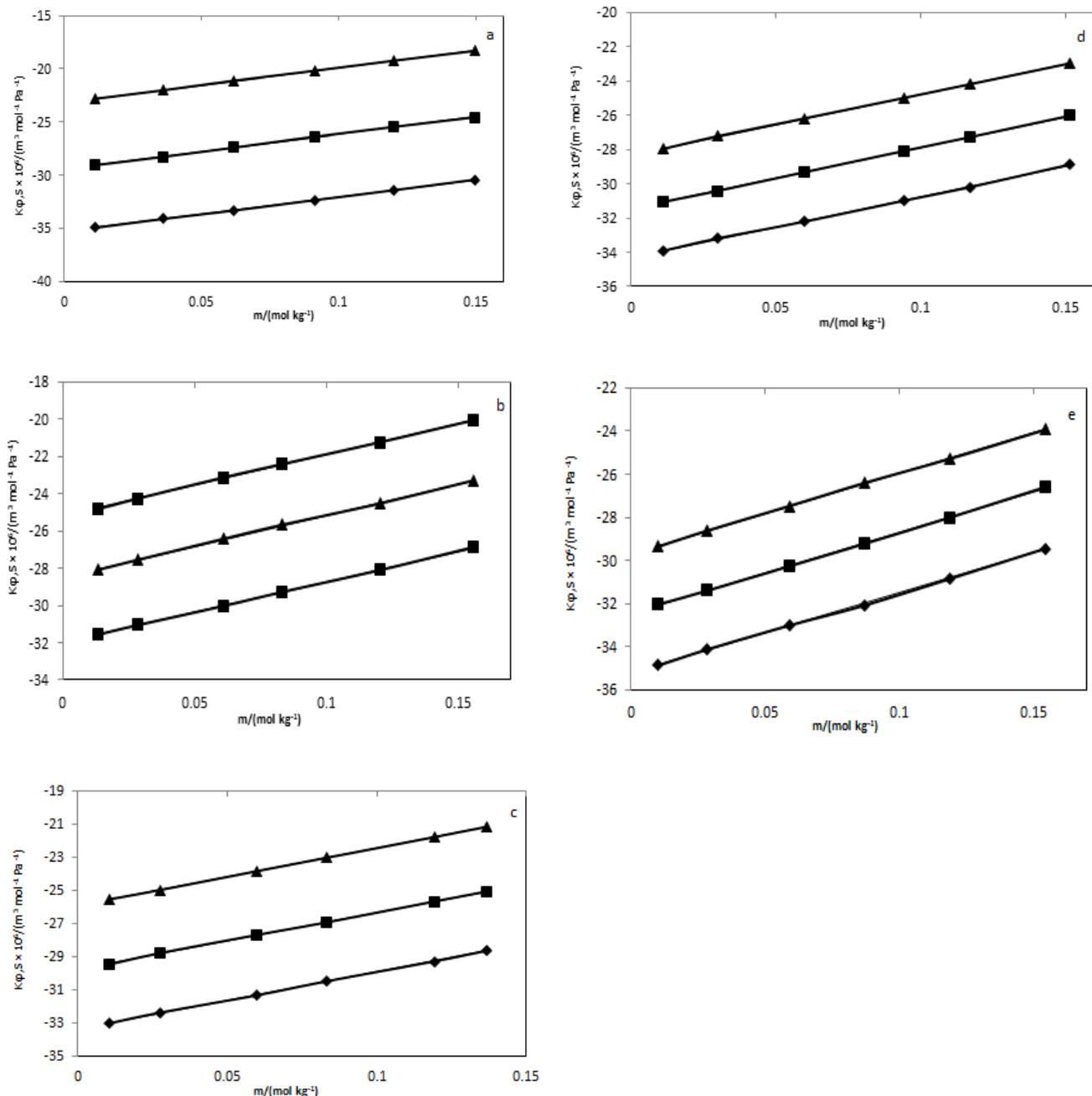
**Fig 2.** Plot of apparent molar volume  $V_\phi$ , vs. molality,  $m$  of (a) histidine in water, (b) histidine in 0.0006 M aqueous nortriptyline, (c) histidine in 0.001 M aqueous nortriptyline, (d) histidine in 0.002 M aqueous nortriptyline, and (e) histidine in 0.004 M aqueous nortriptyline at temperatures: 305.15 K( $\diamond$ ), 310.15 K( $\square$ ), and 315.15 K( $\Delta$ )



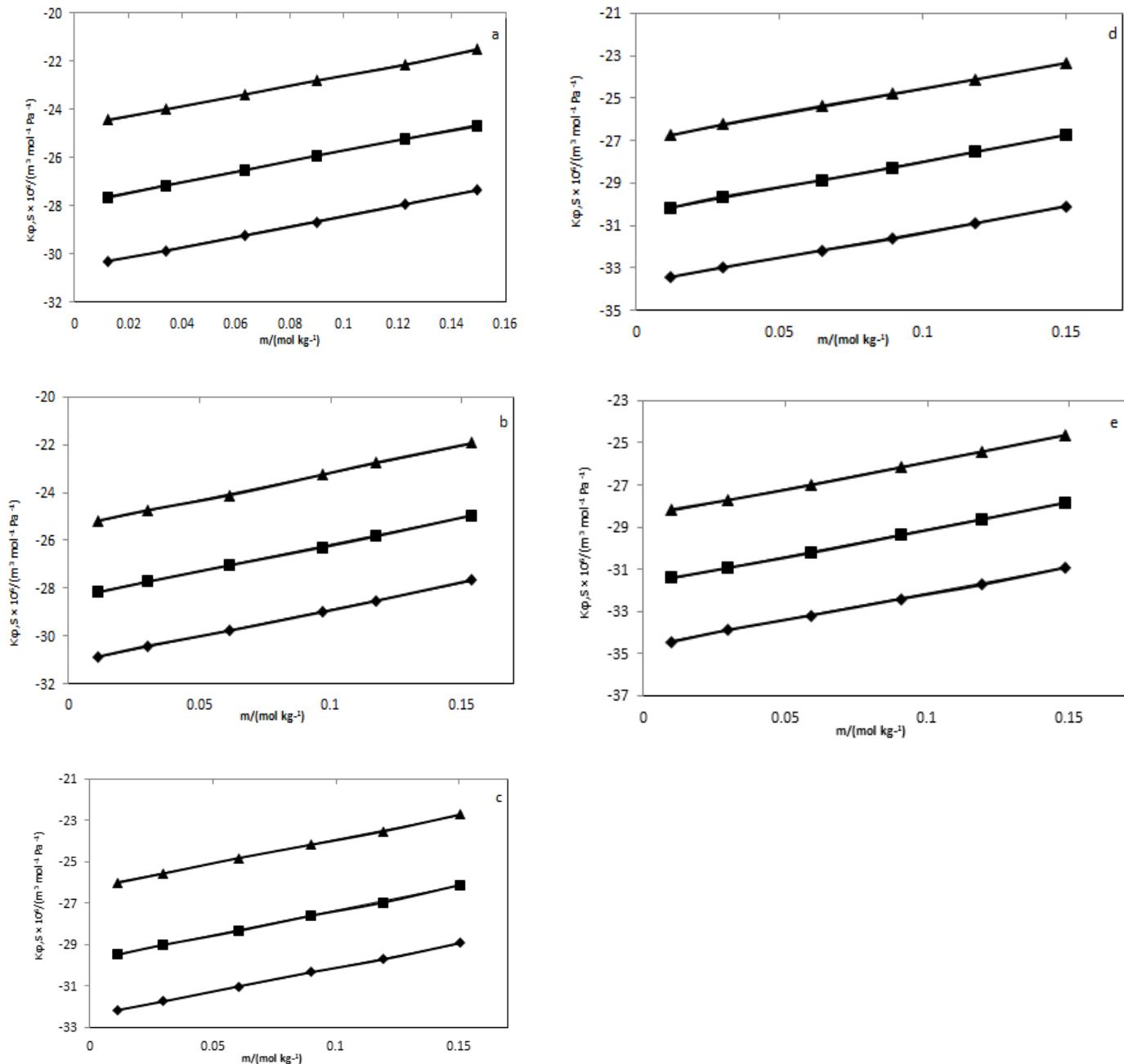
**Fig 3.** Plot of apparent molar volume  $V_\phi$ , vs. molality,  $m$  of (a) valine in water, (b) valine in 0.0006  $m$  aqueous nortriptyline, (c) valine in 0.001  $m$  aqueous nortriptyline, (d) valine in 0.002  $m$  aqueous nortriptyline, and (e) valine in 0.004  $m$  aqueous nortriptyline at temperatures: 305.15 K( $\diamond$ ), 310.15 K( $\square$ ), and 315.15 K( $\Delta$ )



**Fig 4.** Plot of isentropic compressibility  $\kappa_{\phi s}$  vs. molality,  $m$  of (a) glycine in water, (b) glycine in 0.0006  $m$  aqueous nortriptyline, (c) glycine in 0.001  $m$  aqueous nortriptyline, (d) glycine in 0.002  $m$  aqueous nortriptyline, and (e) glycine in 0.004  $m$  aqueous nortriptyline at temperatures: 305.15 K( $\diamond$ ), 310.15 K( $\square$ ), and 315.15 K( $\Delta$ )



**Fig 5.** Plot of isentropic compressibility  $K_{\omega s}$ , vs. molality,  $m$  of (a) histidine in water, (b) histidine in 0.0006  $m$  aqueous nortriptyline, (c) histidine in 0.001  $m$  aqueous nortriptyline, (d) histidine in 0.002  $m$  aqueous nortriptyline, and (e) histidine in 0.004  $m$  aqueous nortriptyline at temperatures: 305.15 K( $\diamond$ ), 310.15 K( $\square$ ), and 315.15 K( $\Delta$ )



**Fig 6.** Plot of isentropic compressibility  $k_{\phi,s}$  vs. molality,  $m$  of (a) valine in water, (b) valine in 0.0006  $m$  aqueous amitriptyline, (c) valine in 0.001  $m$  aqueous amitriptyline, (d) valine in 0.002  $m$  aqueous amitriptyline, and (e) valine in 0.004  $m$  aqueous amitriptyline at temperatures: 305.15 K( $\diamond$ ), 310.15 K( $\square$ ), and 315.15 K( $\Delta$ )

**Table 1.** Specification of chemical samples

Chemical name	Structure	Mass Fraction purity
Nortriptyline		>99%
Valine		>99%
Histidine		>99%
Glycine		>99%

**Table 2.** Densities  $\rho$  and speed of sounds  $u$  of amino acids in water and aqueous nortriptyline solutions at different temperatures

$m/(mol\ Kg^{-1})$	$\rho \times 10^{-3} /$	(kg m $^{-3}$ )			$u/(m\ s^{-1})$		
		$T =$ 305.15 k	$T =$ 310.15 k	$T =$ 315.15 k	$T =$ 305.15 k	$T =$ 310.15 k	$T =$ 315.15 k
Valine+water							
0	0.99504	0.99334	0.99145		1513.7	1523.7	1532.2
0.01	0.99538	0.99367	0.99177		1515.0	1524.9	1533.3
0.03	0.99596	0.99424	0.99233		1517.1	1526.9	1535.3
0.06	0.99674	0.99501	0.99307		1519.9	1529.7	1537.9
0.08	0.99745	0.99570	0.99375		1522.4	1532.1	1540.3
0.12	0.99831	0.99654	0.99457		1525.5	1535.1	1543.1
0.14	0.99901	0.99722	0.99523		1528.0	1537.5	1545.4
Valine+0.0006 m							
Nortriptylin							
0	0.99505	0.99335	0.99145		1513.79	1523.82	1532.27
0.01	0.99536	0.99365	0.99175		1514.92	1524.92	1533.33
0.03	0.99588	0.99416	0.99224		1516.8	1526.75	1535.09
0.06	0.99671	0.99497	0.99304		1519.86	1529.72	1537.95
0.09	0.99765	0.99589	0.99394		1523.26	1533.02	1541.1
0.11	0.99818	0.99641	0.99445		1525.2	1534.9	1542.89
0.15	0.99912	0.99733	0.99534		1528.54	1538.13	1545.99
Valine+0.001 m							
nortriptyline							
0	0.99506	0.99337	0.99146		1513.87	1523.88	1532.32
0.01	0.99538	0.99367	0.99176		1515.05	1525.03	1533.42
0.03	0.99588	0.99416	0.99224		1516.92	1526.85	1535.16
0.07	0.99669	0.99496	0.99302		1519.97	1529.82	1537.99

0.10	0.99747	0.99572	0.99376	1522.88	1532.64	1540.69
0.14	0.99823	0.99647	0.99449	1525.71	1535.38	1543.3
0.17	0.99904	0.99725	0.99525	1528.64	1538.2	1545.99
<b>Valine+0.002 m</b>						
<b>nortriptyline</b>						
0	0.99509	0.99338	0.99147	1513.98	1523.99	1532.49
0.01	0.99541	0.99369	0.99178	1515.21	1525.18	1533.57
0.03	0.99592	0.99418	0.99226	1517.14	1527.04	1535.35
0.06	0.99684	0.99508	0.99314	1520.68	1530.45	1538.6
0.08	0.99748	0.99571	0.99375	1523.14	1532.81	1540.85
0.11	0.99823	0.99644	0.99446	1525.96	1535.51	1543.43
0.14	0.99905	0.99724	0.99525	1529.03	1538.45	1546.23
<b>Valine+0.004 m</b>						
<b>nortriptyline</b>						
0	0.99517	0.99348	0.99151	1514.34	1524.32	1532.66
0.01	0.99543	0.99374	0.99176	1515.4	1525.34	1533.68
0.02	0.99597	0.99426	0.99228	1517.52	1527.4	1535.66
0.06	0.99675	0.99502	0.99302	1520.55	1530.34	1538.48
0.08	0.99758	0.99583	0.99382	1523.78	1533.46	1541.47
0.11	0.99832	0.99655	0.99452	1526.64	1536.21	1544.1
0.14	0.99907	0.99728	0.99524	1529.49	1538.96	1546.73
<b>Glycine+ water</b>						
0	0.99504	0.99334	0.99145	1513.7	1523.7	1532.2
0.01	0.99545	0.99375	0.99185	1514.4	1524.3	1532.8
0.02	0.99601	0.99430	0.99239	1515.3	1525.2	1533.5
0.06	0.99708	0.99536	0.99344	1517.0	1526.8	1535.0
0.08	0.99797	0.99624	0.99431	1518.4	1528.1	1536.2
0.12	0.99907	0.99732	0.99537	1520.0	1529.6	1537.7
0.14	1.00004	0.99828	0.99632	1521.5	1531.0	1538.9
<b>Glycine+0.0006 m</b>						
<b>nortriptyline</b>						
0	0.99505	0.99336	0.99146	1514.01	1523.8	1532.28
0.01	0.99548	0.99379	0.99189	1514.72	1524.48	1532.93
0.03	0.99633	0.99463	0.99272	1516.08	1525.78	1534.17
0.06	0.99707	0.99535	0.99344	1517.24	1526.89	1535.23
0.08	0.99795	0.99622	0.99429	1518.6	1528.2	1536.47
0.12	0.99945	0.99771	0.99576	1520.89	1530.38	1538.53
0.15	1.00015	0.99839	0.99644	1521.92	1531.36	1539.45
<b>Glycine+0.001 m</b>						
<b>nortriptyline</b>						
0	0.99507	0.99337	0.99147	1514.11	1523.89	1532.3
0.01	0.99553	0.99382	0.99192	1514.86	1524.62	1533.01
0.03	0.99617	0.99445	0.99254	1515.9	1525.63	1533.99
0.06	0.99721	0.99549	0.99357	1517.57	1527.25	1535.56
0.09	0.99802	0.99628	0.99435	1518.83	1528.47	1536.74
0.13	0.99916	0.99741	0.99547	1520.58	1530.16	1538.38
0.16	1.00008	0.99832	0.99637	1521.96	1531.49	1539.67
<b>Glycine+0.002 m</b>						

nortriptyline						
0	0.99510	0.99340	0.99150	1514.09	1524.04	1532.49
0.01	0.99546	0.99377	0.99186	1514.72	1524.64	1533.06
0.03	0.99620	0.99450	0.99258	1515.98	1525.83	1534.19
0.06	0.99721	0.99550	0.99357	1517.66	1527.42	1535.69
0.09	0.99816	0.99643	0.99449	1519.2	1528.87	1537.06
0.11	0.99922	0.99748	0.99553	1520.9	1530.46	1538.55
0.15	1.00002	0.99827	0.99631	1522.13	1531.62	1539.63
Glycine+0.004 m						
nortriptyline						
0	0.99515	0.99342	0.99156	1514.41	1524.35	1532.75
0.01	0.99555	0.99382	0.99195	1515.11	1525.03	1533.41
0.03	0.99618	0.99444	0.99257	1516.19	1526.08	1534.43
0.06	0.99729	0.99554	0.99365	1518.06	1527.9	1536.19
0.09	0.99835	0.99658	0.99469	1519.8	1529.58	1537.82
0.12	0.99935	0.99757	0.99566	1521.39	1531.11	1539.29
0.16	1.00067	0.99888	0.99695	1523.41	1533.07	1541.17
Histidine+ water						
0	0.99504	0.99334	0.99145	1513.7	1523.7	1532.2
0.01	0.99566	0.99396	0.99206	1514.8	1524.6	1533.5
0.03	0.99704	0.99534	0.99344	1517.0	1526.7	1535.2
0.06	0.99848	0.99677	0.99486	1519.2	1528.7	1537.0
0.09	1.00013	0.99841	0.99649	1521.7	1530.9	1538.9
0.11	1.00174	1.00002	0.99809	1524.0	1532.9	1540.6
0.14	1.00340	1.00168	0.99974	1526.3	1534.9	1542.3
Histidine+0.0006						
m nortriptyline						
0	0.99506	0.99335	0.99146	1513.9	1523.88	1532.3
0.01	0.99580	0.99410	0.99219	1515.06	1524.98	1533.34
0.02	0.99664	0.99493	0.99302	1516.34	1526.19	1534.48
0.05	0.99845	0.99672	0.99480	1519	1528.69	1536.83
0.08	0.99969	0.99796	0.99602	1520.75	1530.33	1538.37
0.11	1.00177	1.00002	0.99807	1523.55	1532.95	1540.81
0.14	1.00375	1.00198	1.00002	1526.04	1535.26	1542.96
Histidine+0.001						
m nortriptyline						
0	0.99507	0.99338	0.99147	1513.92	1523.9	1532.34
0.01	0.99567	0.99397	0.99206	1514.88	1524.81	1533.27
0.02	0.99661	0.99491	0.99299	1516.36	1526.21	1534.58
0.05	0.99839	0.99668	0.99474	1519.07	1528.76	1536.95
0.08	0.99970	0.99797	0.99603	1520.96	1530.55	1538.6
0.12	1.00171	0.99997	0.99801	1523.75	1533.15	1540.99
0.15	1.00269	1.00094	0.99898	1525.03	1534.35	1542.08
Histidine+0.002						
m nortriptyline						
0	0.995083	0.993402	0.991499	1513.99	1524.02	1532.47
0.01	0.995699	0.994014	0.992105	1515	1525	1533.4
0.02	0.996742	0.99505	0.993131	1516.67	1526.61	1534.93

0.05	0.998395	0.996691	0.994756	1519.23	1529.06	1537.26
0.08	1.000318	0.998601	0.996649	1522.06	1531.76	1539.82
0.11	1.001589	0.999864	0.997903	1523.85	1533.46	1541.42
0.15	1.003509	1.001774	0.999796	1526.39	1535.88	1543.7
<b>Histidine+0.004 m nortriptyline</b>						
0	0.99516	0.99345	0.99155	1514.45	1524.4	1532.8
0.009	0.99575	0.99401	0.99211	1515.37	1525.32	1533.69
0.02	0.99676	0.99501	0.99309	1517.02	1526.92	1535.23
0.05	0.99846	0.99669	0.99475	1519.69	1529.5	1537.71
0.08	1.00003	0.99824	0.99628	1522.06	1531.77	1539.88
0.12	1.0018	0.99999	0.99801	1524.57	1534.18	1542.2
0.14	1.00379	1.00196	0.99995	1527.22	1536.7	1544.62

**Table 3.** Apparent molar volumes  $V_\varphi$  and isentropic compression  $V_{\varphi,s}$  of amino acids in water and aqueous nortriptyline solutions at different temperatures

$m/(mol\ Kg^{-1})$	$V_\varphi \times 10^{-6} / (\text{m}^3 \text{ mol}^{-1})$			$K_{\varphi,s} \times 10^6 / (\text{m}^3 \text{ mol}^{-1} \text{ Pa}^{-1})$		
	$T =$	$T =$	$T =$	$T =$	$T =$	$T =$
	305.15 k	310.15 k	315.15 k	305.15 k	310.15 k	315.15 k
<b>Valine+water</b>						
0.01	90.04	90.80	91.57	-30.28	-27.65	-24.43
0.03	90.13	90.86	91.64	-29.85	-27.15	-24
0.06	90.22	90.97	91.76	-29.23	-26.52	-23.39
0.08	90.32	91.07	91.86	-28.66	-25.92	-22.81
0.12	90.45	91.20	91.98	-27.94	-25.23	-22.16
0.14	90.53	91.30	92.07	-27.34	-24.69	-21.50
<b>Valine+0.0006 m nortriptylin</b>						
0.01	90.13	90.87	91.62	-30.86	-28.17	-25.17
0.03	90.20	90.94	91.67	-30.41	-27.72	-24.73
0.06	90.30	91.04	91.77	-29.77	-27.04	-24.10
0.09	90.42	91.16	91.88	-28.99	-26.28	-23.24
0.11	90.50	91.22	91.95	-28.52	-25.81	-22.73
0.15	90.62	91.34	92.07	-27.65	-24.94	-21.92
<b>Valine+0.001 m nortriptyline</b>						
0.009	90.17	90.89	91.63	-32.15	-29.48	-26.00
0.03	90.22	90.97	91.71	-31.74	-29.01	-25.56
0.06	90.35	91.08	91.84	-31.01	-28.33	-24.81
0.08	90.46	91.19	91.96	-30.33	-27.59	-24.16
0.11	90.57	91.30	92.08	-29.71	-26.96	-23.52
0.14	90.68	91.410	92.20	-28.91	-26.10	-22.70
<b>Valine+0.002 m nortriptyline</b>						
0.01	90.20	91.00	91.73	-33.43	-30.16	-26.76

0.03	90.27	91.05	91.77	-32.97	-29.64	-26.23
0.06	90.42	91.17	91.90	-32.16	-28.85	-25.36
0.08	90.51	91.26	91.99	-31.62	-28.27	-24.79
0.11	90.62	91.35	92.08	-30.90	-27.53	-24.11
0.14	90.74	91.46	92.19	-30.08	-26.72	-23.33
Valine+0.004 m						
nortriptyline						
0.01	90.34	91.07	91.72	-34.44	-31.40	-28.15
0.02	90.42	91.14	91.81	-33.87	-30.92	-27.70
0.06	90.51	91.26	91.92	-33.18	-30.21	-26.99
0.08	90.62	91.37	92.05	-32.40	-29.37	-26.16
0.11	90.71	91.47	92.15	-31.72	-28.62	-25.41
0.14	90.82	91.57	92.27	-30.91	-27.83	-24.63
Glycine+ water						
0.01	41.11	41.53	41.96	-29.04	-26.37	-23.83
0.02	41.14	41.58	42.02	-28.71	-26.11	-23.47
0.06	41.23	41.66	42.11	-28.20	-25.53	-22.91
0.08	41.30	41.74	42.19	-27.76	-25.08	-22.43
0.12	41.39	41.83	42.28	-27.18	-24.52	-21.87
0.14	41.46	41.90	42.36	-26.65	-23.99	-21.33
Glycine+0.0006						
m nortriptyline						
0.01	41.18	41.58	41.99	-28.96	-26.88	-24.94
0.02	41.24	41.66	42.07	-28.51	-26.40	-24.43
0.06	41.30	41.72	42.14	-28.09	-26.00	-24.05
0.08	41.36	41.79	42.21	-27.60	-25.59	-23.59
0.12	41.47	41.91	42.35	-26.86	-24.81	-22.77
0.15	41.53	41.97	42.41	-26.50	-24.45	-22.39
Glycine+0.001 m						
nortriptyline						
0.01	41.26	41.65	42.04	-29.11	-27.48	-26.02
0.02	41.32	41.71	42.10	-28.75	-27.11	-25.62
0.06	41.41	41.79	42.19	-28.19	-26.56	-25.06
0.09	41.47	41.85	42.25	-27.75	-26.11	-24.60
0.13	41.57	41.95	42.34	-27.11	-25.47	-23.99
0.16	41.65	42.02	42.42	-26.60	-24.96	-23.51
Glycine+0.002 m						
nortriptyline						
0.01	41.29	41.67	42.15	-30.65	-28.21	-25.92
0.03	41.35	41.74	42.22	-30.20	-27.66	-25.41
0.05	41.44	41.84	42.29	-29.57	-27.07	-24.78
0.09	41.51	41.93	42.35	-28.99	-26.46	-24.22
0.12	41.61	42.02	42.43	-28.36	-25.81	-23.55
0.14	41.69	42.08	42.49	-27.79	-25.29	-23.02
Glycine+0.004 m						
nortriptyline						
0.01	41.35	41.79	42.23	-30.85	-29.03	-27.46
0.03	41.41	41.82	42.27	-30.44	-28.69	-27.10

0.06	41.49	41.90	42.34	-29.74	-28.05	-26.41
0.09	41.57	41.97	42.42	-29.10	-27.38	-25.78
0.12	41.65	42.04	42.48	-28.46	-26.72	-25.09
0.16	41.75	42.13	42.56	-27.56	-25.88	-24.24
Histidine+ water						
0.01	99.47	99.73	100.08	-34.95	-29.06	-22.81
0.03	99.31	99.58	99.92	-34.11	-28.28	-22.02
0.06	99.13	99.40	99.77	-33.33	-27.41	-21.12
0.09	98.95	99.24	99.59	-32.35	-26.38	-20.18
0.11	98.77	99.05	99.40	-31.42	-25.46	-19.24
0.14	98.59	98.87	99.22	-30.42	-24.55	-18.31
Histidine+0.0006						
<i>m</i> nortriptyline						
0.01	99.53	99.98	100.52	-31.55	-28.07	-24.81
0.02	99.43	99.90	100.41	-31.04	-27.53	-24.24
0.05	99.20	99.70	100.21	-30.02	-26.41	-23.12
0.08	99.06	99.55	100.06	-29.27	-25.66	-22.40
0.11	98.83	99.32	99.84	-28.09	-24.51	-21.22
0.14	98.61	99.11	99.61	-26.86	-23.27	-20.04
Histidine+0.001						
<i>m</i> nortriptyline						
0.01	99.63	100.08	100.64	-33.01	-29.45	-25.52
0.02	99.52	100.00	100.52	-32.36	-28.77	-24.97
0.05	99.32	99.78	100.31	-31.33	-27.68	-23.84
0.08	99.14	99.64	100.16	-30.48	-26.92	-23.03
0.12	98.92	99.41	99.92	-29.28	-25.66	-21.77
0.15	98.81	99.30	99.80	-28.64	-25.06	-21.15
Histidine+0.002						
<i>m</i> nortriptyline						
0.01	99.71	100.15	100.79	-33.90	-31.05	-27.95
0.02	99.60	100.05	100.68	-33.18	-30.40	-27.21
0.05	99.40	99.87	100.50	-32.21	-29.32	-26.18
0.08	99.19	99.65	100.27	-30.98	-28.07	-24.98
0.11	99.06	99.52	100.11	-30.18	-27.26	-24.17
0.15	98.86	99.30	99.88	-28.86	-26.01	-22.96
Histidine+0.004						
<i>m</i> nortriptyline						
0.009	99.73	100.40	101.18	-34.86	-32.03	-29.34
0.02	99.61	100.29	101.05	-34.12	-31.39	-28.61
0.05	99.40	100.10	100.87	-33.01	-30.26	-27.47
0.08	99.21	99.92	100.69	-32.07	-29.21	-26.39
0.12	99.00	99.72	100.49	-30.84	-28.00	-25.28
0.14	98.78	99.47	100.26	-29.43	-26.58	-23.92

**Table 4.** Limiting apparent molar volume  $V_\omega^*$ , limiting apparent molar isentropic compression  $\mathbf{k}^0_{\omega,s}$ ,  $S_v$ , and  $S_k$  aqueous nortriptyline solutions at different temperatures

$m/(mol \ kg^{-1})$	$T/(k)$	$kg^{-1}$ ( $m_3 mol^{-1}$ )	$S_v \times 10^6$ ( $m^3 mol^{-1} kg^{-1}$ )	$\mathbf{k}^0_{\omega,s} \times 10^6$ ( $m_3 mol^{-1} pa^{-1}$ )	$S_k \times 10^6$ ( $m_3 mol^{-1} pa^{-1} kg^{-1}$ )
Valine					
0.0006	305.15	90.09	3.77	-31.12	22.3
	310.15	90.83	3.55	-28.42	22.39
	315.15	91.58	3.60	-25.44	22.88
0.001	305.15	90.12	3.79	-32.43	23.18
	310.15	90.86	3.69	-29.76	23.94
	315.15	91.59	4.11	-26.26	23.41
0.002	305.15	90.15	3.93	-33.72	24
	310.15	90.95	3.42	-30.43	24.59
	315.15	91.68	3.42	-27	24.54
0.004	305.15	90.37	3.85	-34.66	25.06
	310.15	91.06	3.63	-31.69	25.78
	315.15	91.80	3.75	-28.45	25.57
Glycine					
0.0006	305.15	41.15	2.44	-29.17	17.58
	310.15	41.55	2.75	-27.07	17.19
	315.15	41.96	2.93	-25.14	18.01
0.001	305.15	41.23	2.72	-29.35	18.29
	310.15	41.61	2.79	-27.72	18.35
	315.15	42.01	2.77	-26.23	18.24
0.002	305.15	41.26	2.89	-30.88	20.60
	310.15	41.64	3.04	-28.40	21.01
	315.15	42.13	2.39	-26.12	20.87
0.004	305.15	41.33	2.54	-31.10	21.11
	310.15	41.76	2.23	-29.32	20.48
	315.15	42.20	2.18	-27.74	20.88
Histidine					
0.0006	305.15	99.61	-6.45	-31.99	32.79
	310.15	100.07	-6.16	-28.49	33.48
	315.15	100.6	-6.33	-25.2	33.25
0.001	305.15	99.7	-6.54	-33.36	34.41
	310.15	100.16	-6.26	-29.77	34.54
	315.15	100.71	-6.60	-25.91	34.79
0.002	305.15	99.77	-6.08	-34.29	35.52
	310.15	100.23	-6.09	-31.46	35.98
	315.15	100.86	-6.46	-28.30	35.36
0.004	305.15	99.79	-6.61	-35.22	37.14
	310.15	100.48	-6.36	-32.46	37.75
	315.15	101.25	-6.35	-29.69	37.39

**Table 5.** Partial molar volume of transfer  $V_{\phi,tr}$ , and partial molar isentropic compression of transfer  $k_{\phi,s,tr}$ , of amino acids in water and in aqueous nortriptyline solutions at different temperatures

$m/(mol \text{ kg}^{-1})$	$V_{\phi,tr} \times 10^6 / (\text{m}^3 \text{ mol}^{-1})$			$k_{\phi,s,tr} \times 10^6 / (\text{m}^3 \text{ mol}^{-1} \text{ pa}^{-1})$		
	$T = 305.15 \text{ k}$	$T = 310.15 \text{ k}$	$T = 315.15 \text{ k}$	$T = 305.15 \text{ k}$	$T = 310.15 \text{ k}$	$T = 315.15 \text{ k}$
Valine						
0.0006	0.08	0.09	0.06	0.54	0.62	0.65
0.001	0.11	0.11	0.07	1.85	1.86	1.54
0.002	0.15	0.20	0.16	3.14	2.64	2.28
0.004	0.31	0.29	0.16	4.09	3.80	3.74
Histidine						
0.0006	0.07	0.27	0.45	0.37	1.12	2.02
0.001	0.16	0.36	0.56	1.74	2.41	2.74
0.002	0.23	0.43	0.72	2.67	4.09	5.13
0.004	0.25	0.67	1.10	3.60	5.09	6.52
Glycine						
0.0006	0.08	0.04	0.02	0.39	0.47	1.13
0.001	0.15	0.11	0.06	0.58	1.13	2.22
0.002	0.18	0.13	0.19	2.10	1.80	2.11
0.004	0.25	0.25	0.26	2.32	2.72	2.72

**References**

- [1] Nain AK, Pal R, Sharma RK. Volumetric, ultrasonic, and viscometric behaviour of L-histidine in aqueous-glucose solutions at different temperatures, *The Journal of Chemical Thermodynamics*; 2011 Apr 1; 43(4):603-12.  
<https://doi.org/10.1016/j.jct.2010.11.017>
- [2] Nain AK, Pal R, Sharma RK. Physicochemical study of solute-solute and solute-solvent interactions of L-histidine in water+ sucrose solutions at different temperatures, *Journal of Molecular Liquids*; 2012 Jan 1; 165:154-60.  
<https://doi.org/10.1016/j.molliq.2011.11.003>
- [3] Pal A, Chauhan N. Densities, speeds of sound and viscosities of L-alanine in aqueous fructose, maltose and lactose solutions at different temperatures.
- [4] Kulikova GA, Parfenyuk EV. Influence of side chain of L- $\alpha$ -amino acids on their interaction with D-glucose in dilute aqueous solutions, *Journal of solution chemistry*; 2008 Jun; 37:835-40. <https://doi.org/10.1007/s10953-008-9275-1>
- [5] Iqbal MJ, Chaudhry MA. Thermodynamic study of three pharmacologically significant drugs: Density, viscosity, and refractive index measurements at different temperatures, *The Journal of Chemical Thermodynamics*; 2009 Feb 1; 41(2):221-6.  
<https://doi.org/10.1016/j.jct.2008.09.016>
- [6] Dhondge SS, Zodape SP, Parwate DV. Volumetric and viscometric studies of some drugs in aqueous solutions at different temperatures, *The Journal of Chemical Thermodynamics*; 2012 May 1; 48:207-12.  
<https://doi.org/10.1016/j.jct.2011.12.022>
- [7] Banipal TS, Kaur J, Banipal PK, Singh K. Study of interactions between amino acids and zinc chloride in aqueous solutions through volumetric measurements at T=(288.15 to 318.15) K, *Journal of Chemical & Engineering Data*; 2008 Aug 14; 53(8):1803-16.  
<https://doi.org/10.1021/je8001464>

- [8] Ali A, Bidhuri P, Uzair S. Thermodynamic and spectroscopic studies of alanine and phenylalanine in aqueous  $\beta$ -cyclodextrin solutions, *Journal of Saudi Chemical Society*; 2017 Jan 1; 21:S136-42.  
<https://doi.org/10.1016/j.jscs.2013.11.003>
- [9] Kumar M, Sawhney N, Sharma AK, Sharma M. Volumetric, acoustic and viscometric studies of L-histidine in aqueous solutions of non-steroid anti-inflammatory drug ketorolac tromethamine at different temperatures, *Journal of Molecular Liquids*; 2017 Oct 1; 243:41-51.  
<https://doi.org/10.1016/j.molliq.2017.08.001>
- [10] Gillman PK. Tricyclic antidepressant pharmacology and therapeutic drug interactions updated, *British journal of pharmacology*; 2007 Jul; 151(6):737-48.  
<https://doi.org/10.1038/sj.bjp.0707253>
- [11] Wood SE, Wood EG, Boyd D. *Mastering the world of psychology*, Boston; 2011.
- [12] Singh M, Sharma s, Singh J, Sharma S, Sharma A.M, Sharma S.M, Volumetric, acoustic and viscometric studies of solute-solute and solute-solvent interactions of glycine and its peptides in aqueous solutions of an antidepressant drug at different temperatures, *Journal of Molecular Liquids*; 2022 May, 15; 354:118760.  
<https://doi.org/10.1016/j.molliq.2022.118760>.
- [13] Wadi RK, Ramasami P. Partial molal volumes and adiabatic compressibilities of transfer of glycine and DL-alanine from water to aqueous sodium sulfate at 288.15, 298.15 and 308.15 K. *Journal of the Chemical Society, Faraday Transactions*; 1997; 93(2):243-7.  
<https://doi.org/10.1039/A604650I>
- [14] Soto A, Arce A, Khoshkbarchi MK. Experimental data and modelling of apparent molar volumes, isentropic compressibilities and refractive indices in aqueous solutions of glycine+ NaCl, *Biophysical chemistry*; 1998 Sep 14; 74(3):165-73.
- [https://doi.org/10.1016/S0301-4622\(98\)00183-5](https://doi.org/10.1016/S0301-4622(98)00183-5)
- [15] Sharma SK, Singh G, Kumar H, Kataria R. Solvation behavior of some amino acids in aqueous solutions of non-steroidal anti-inflammatory drug sodium ibuprofen at different temperatures analysed by volumetric and acoustic methods, *The Journal of Chemical Thermodynamics*; 2016 Jul 1; 98:214-30.  
<https://doi.org/10.1016/j.jct.2016.03.016>
- [16] Yan Z, Wang J, Kong W, Lu J. Effect of temperature on volumetric and viscosity properties of some  $\alpha$ -amino acids in aqueous calcium chloride solutions, *Fluid Phase Equilibria*; 2004 Feb 1; 215(2):143-50.  
<https://doi.org/10.1016/j.fluid.2003.07.001>
- [17] Sawhney N, Kumar M, Sharma AK, Sharma M. Structure making behaviour of L-histidine in aqueous solutions of ampicillin sodium at different temperatures: using physicochemical methods, *The Journal of Chemical Thermodynamics*; 2017 Dec 1; 115:156-70.  
<https://doi.org/10.1016/j.jct.2017.07.040>
- [18] Rajagopal K, Jayabalakrishnan SS. Ultrasonic studies of 4-aminobutyric acid in aqueous metformin hydrochloride solutions at different temperatures, *International Journal of Thermophysics*; 2010 Dec; 31(11-12):2225-38.  
<https://doi.org/10.1007/s10765-010-0862-1>
- [19] Rodríguez H, Soto A, Arce A, Khoshkbarchi MK. Apparent molar volume, isentropic compressibility, refractive index, and viscosity of DL-alanine in aqueous NaCl solutions, *Journal of solution chemistry*; 2003 Jan; 32:53-63.  
<https://doi.org/10.1023/A:1022640715229>
- [20] JKirkwood JG. Theoretical Studies upon Dipolar Ions, *Chemical Reviews*; 1939 Apr 1; 24(2):233-51.  
<https://doi.org/10.1021/cr60078a004>
- [21] Ramasami P, Kakkar R. Partial molar volumes and adiabatic compressibilities at

infinite dilution of aminocarboxylic acids and glycylglycine in water and aqueous solutions of sodium sulphate at (288.15, 298.15 and 308.15) K, *The Journal of Chemical Thermodynamics*; 2006 Nov 1; 38(11):1385-95.

<https://doi.org/10.1016/j.jct.2006.01.014>

[22] Sawhney N, Kumar M, Sharma AK, Sharma M. Thermo-physical properties of l-Alanine/l-Valine in aqueous solutions of non steroid anti inflammatory drug dolonex at different temperatures: Volumetric and acoustic approach, *The Journal of Chemical Thermodynamics*; 2018 Aug 1; 123:22-31.

<https://doi.org/10.1016/j.jct.2018.03.022>

[23] Hedwing, G. R. Thermodynamic properties of peptide solutions 3. Partial molar volumes and partial molar heat capacities of some tripeptides in aqueous solutionJ. *Journal of Solution Chemistry*; 1988 Apr; 17:383-397.  
<https://doi.org/10.1007/BF00650418>

[24] Mathieson JG, Conway BE. Partial molal compressibilities of salts in aqueous solution and assignment of ionic contributions, *Journal of Solution Chemistry*; 1974 Jun; 3:455-77.

<https://doi.org/10.1007/BF00651536>

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